

Original Research Paper

## Combining pieces: a thorough analysis of light activation boosting power and co-substrate preferences for the catalytic efficiency of lytic polysaccharide monoxygenase *MtLPMO9A*

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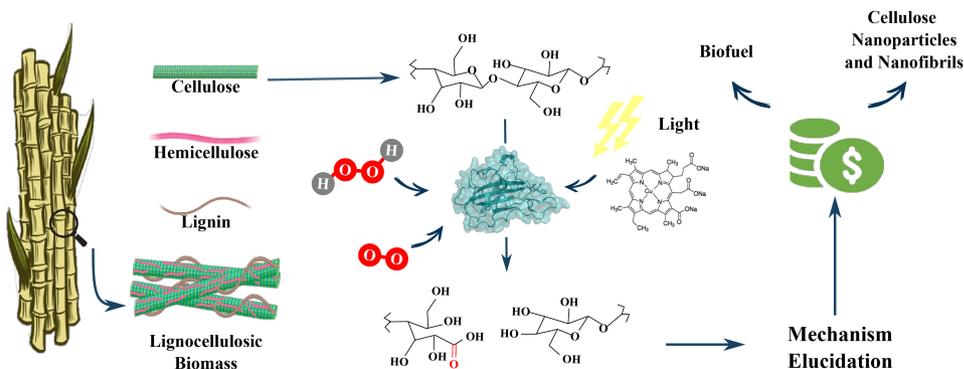
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### HIGHLIGHTS

- *MtLPMO9A* can be efficiently activated by light at the presence of chlorophyllin.
- Photoactivation improves performance of *MtLPMO9A* on crystalline cellulose.
- *MtLPMO9A* can use both O<sub>2</sub> and H<sub>2</sub>O<sub>2</sub> as co-substrates.
- For H<sub>2</sub>O<sub>2</sub> reactions *MtLPMO9A* requires chemical reductant.
- Photobiocatalysis mediated by LPMOs might play a role in plant biomass valorization.

### GRAPHICAL ABSTRACT



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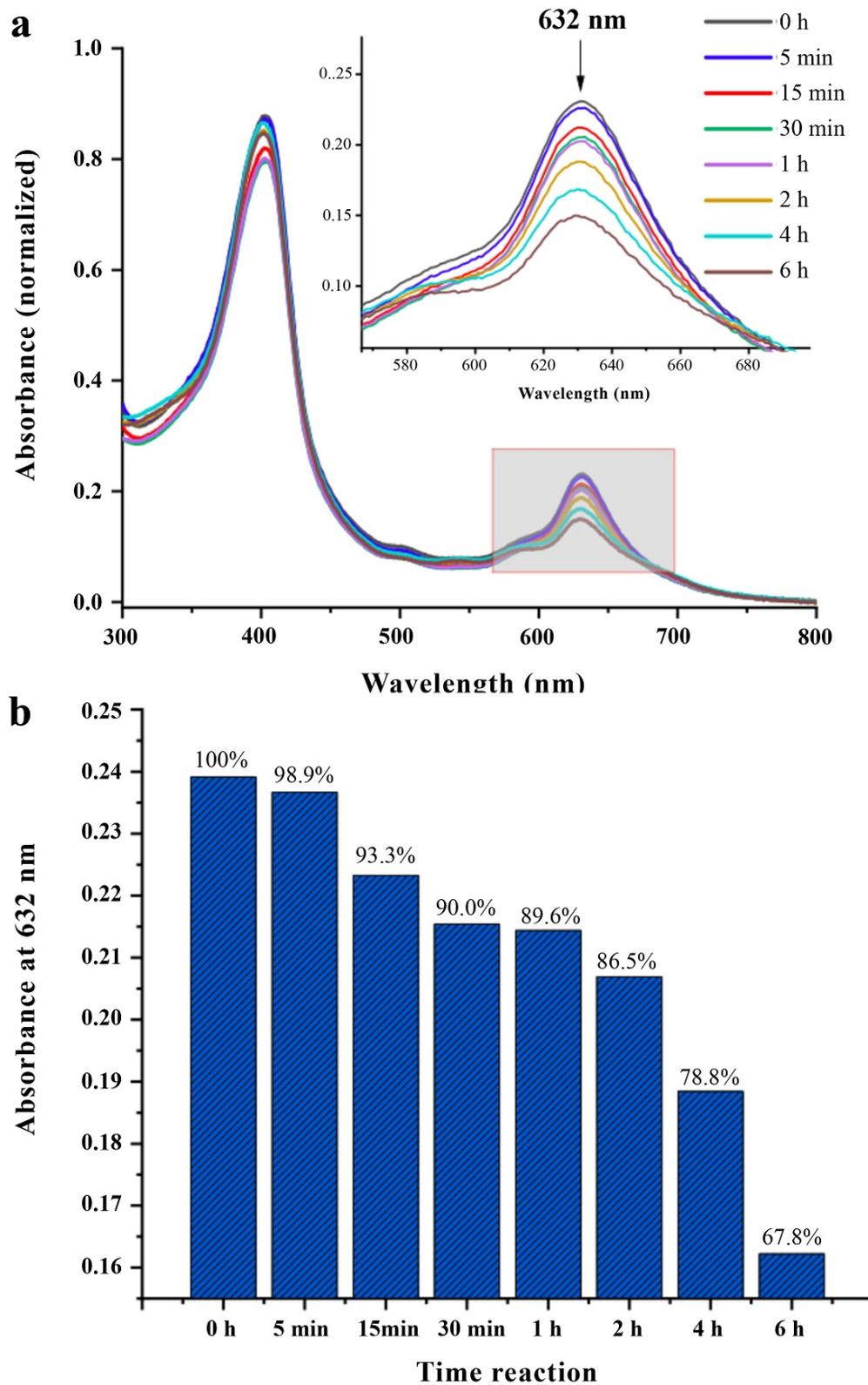
Biomass conversion

### ABSTRACT

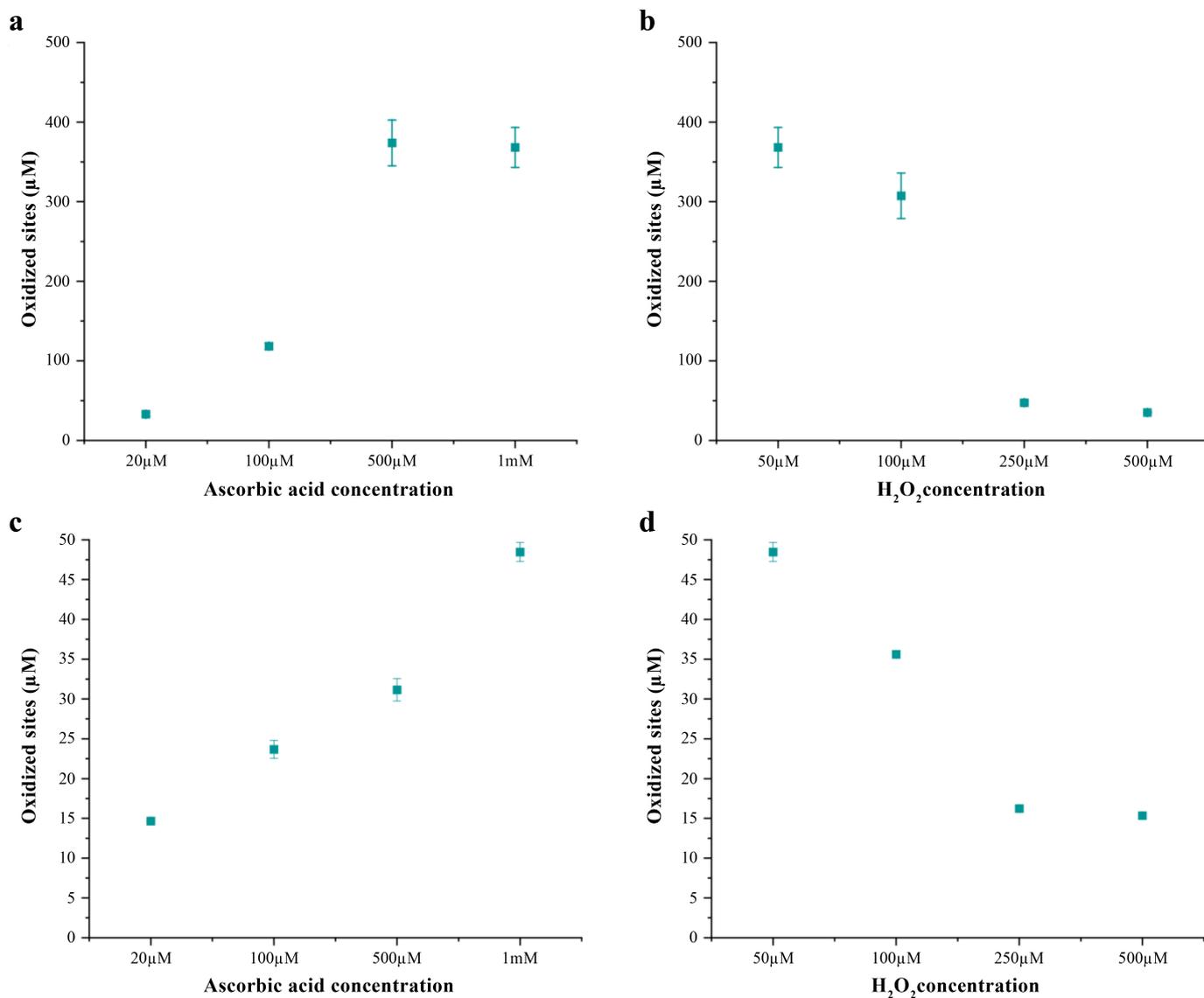
Cost-efficient plant biomass conversion using biochemical and/or chemical routes is essential for transitioning to sustainable chemical technologies and renewable biofuels. Lytic polysaccharide monoxygenases (LPMOs) are copper-dependent enzymes that make part of modern hydrolytic cocktails destined for plant biomass degradation. Here, we characterized *MtLPMO9A* from *Thermothelomyces thermophilus* M77 (formerly *Myceliophthora thermophila*) and demonstrated that it could be efficiently driven by chlorophyllin excited by light in the presence of a reductant agent. However, in the absence of chemical reductant, chlorophyllin and light alone do not lead to a significant release of the reaction products by the LPMO, indicating a low capacity of *MtLPMO9A* reduction (either *via* direct electron transfer or *via* superoxide ion, O<sub>2</sub><sup>-</sup>). We showed that photocatalysis could significantly increase the LPMO activity against highly crystalline and recalcitrant cellulosic substrates, which are poorly degraded in the absence of chlorophyllin and light. We also evaluated the use of co-substrates by *MtLPMO9A*, revealing that the enzyme can use both hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and molecular oxygen (O<sub>2</sub>) as co-substrates for cellulose catalytic oxidation.

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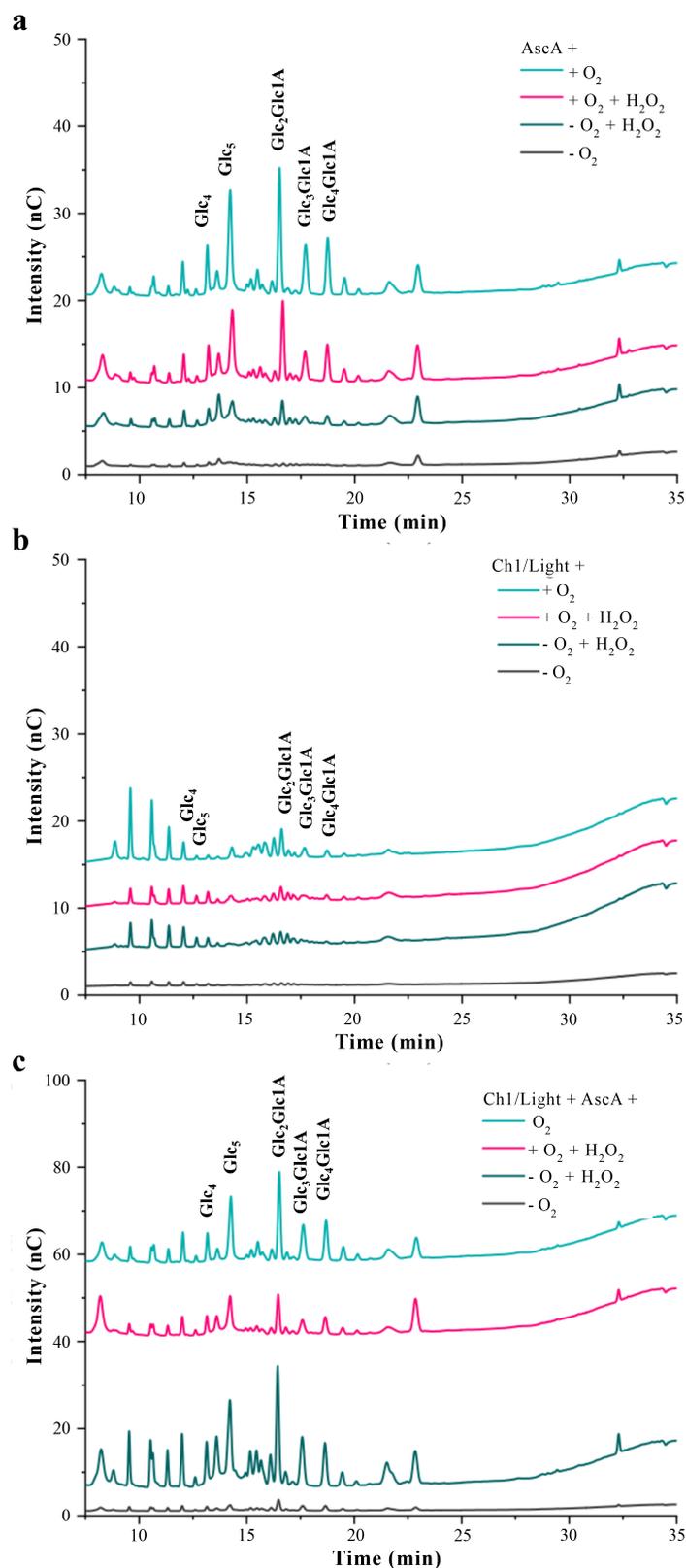
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**Fig. S1.** Assessment of chlorophyllin degradation in reactions over time in the presence of AscA. (a) Evaluation of the behavior of chlorophyllin absorbance spectra over time (632 nm peak) and (b) Quantitative trend of chlorophyllin absorbance over time.



**Fig. S2.** Evaluation of the optimum conditions for peroxygenase reaction of *MtLPMO9A*. Effects of ascorbic acid and H<sub>2</sub>O<sub>2</sub> on *MtLPMO9A* activity on PASC and Avicel were evaluated as follows: (a) reaction was performed with a fixed H<sub>2</sub>O<sub>2</sub> concentration of 50 µM, varying the amount of ascorbic acid from 20 µM to 1 mM on PASC as a substrate; (b) Using PASC as a substrate, ascorbic acid concentration was fixed at 1 mM and the amount of H<sub>2</sub>O<sub>2</sub> was varied from 50 µM to 500 µM. (c) Using Avicel as a substrate, reaction was performed with fixed H<sub>2</sub>O<sub>2</sub> concentration of 50 µM and varying the concentration of ascorbic acid from 20 µM to 1 mM. (d) Using Avicel as a substrate, ascorbic acid concentration was fixed at 1 mM and the concentration of H<sub>2</sub>O<sub>2</sub> was varied from 50 µM to 500 µM. Reactions were performed for 4 h at 50 °C and under constant agitation of 1000 rpm.



**Fig. S3.** Evaluation of *MlLPMO9A* activity under different oxygen/hydrogen peroxide conditions on Avicel. *MlLPMO9A* activity was evaluated on Avicel for 30 min using different combinations of co-substrates: (a) ascorbic acid as an electron donor  $\pm$  O<sub>2</sub>  $\pm$  H<sub>2</sub>O<sub>2</sub>, (b) photoactivated system with 500 $\mu$ M of chlorophyllin  $\pm$  O<sub>2</sub>  $\pm$  H<sub>2</sub>O<sub>2</sub>, and (c) coupled photoactivated system that combines 1mM of ascorbic acid with 500 $\mu$ M of chlorophyllin  $\pm$  O<sub>2</sub>  $\pm$  H<sub>2</sub>O<sub>2</sub>. The highlighted non-oxidized oligosaccharides are Glc<sub>4</sub> (cellotetraose) and Glc<sub>5</sub> (cellopentose). C1-oxidized oligosaccharides are GlcGlc1A (cellobionic acid), Glc<sub>2</sub>Glc1A (cellotronic acid), Glc<sub>3</sub>Glc1A (cellotetraonic acid) and Glc<sub>4</sub>Glc1A (cellopentaonic acid). Peak assignment was done using standards.